

ERYTHRODIENE: A NEW SPIROBICYCLIC SESQUITERPENE OF A RARE SKELETAL CLASS
FROM THE CARIBBEAN GORGONIAN CORAL *ERYTHROPODIUM CARIBAEORUM*

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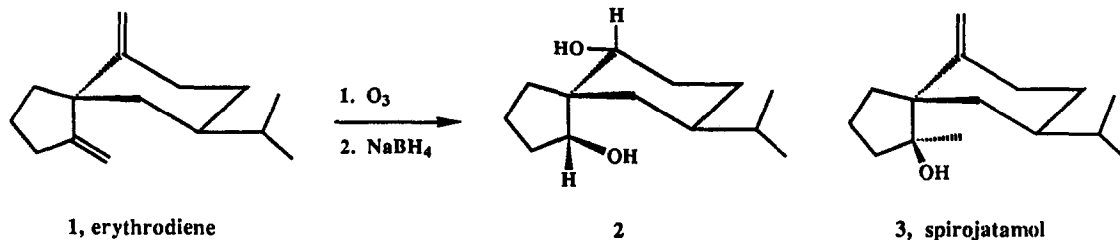
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Summary - Erythrodiene (1), a new sesquiterpene hydrocarbon of a rare structural class, has been isolated from the encrusting Caribbean gorgonian coral *Erythropodium caribaeorum*. The structure of the new compound was assigned on the basis of spectral studies and through an X-ray analysis of a reduced ozonolysis product, the diol 2.

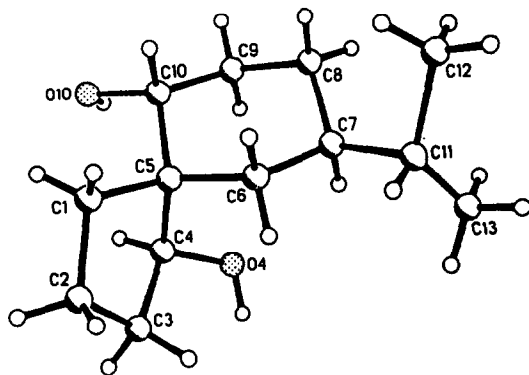
The Caribbean gorgonian octocoral *Erythropodium caribaeorum* (Duchassaing & Michelloti) grows aggressively, forming thin sheets over hard corals to gain substrate.^{1a} Despite their exposure and their lack of a hard coral's calcareous exoskeleton, these animals are relatively free of predation.^{1b} As part of our interest in the chemical adaptations of this group of soft-corals, we recently investigated the chemical basis for predator avoidance of *E. caribaeorum*.^{1c} The majority of the deterrence noted toward predatory reef fishes was derived from a mixture containing two chlorine-containing diterpenoids, erythrolides B and D. While the non-polar fractions were inactive, it was noted that a single sesquiterpene hydrocarbon, erythrodiene (1), comprised over 60% of these mixtures. We now wish to report the structure determination of this compound as a member of a rare spirobicyclo[5.4]decane terpenoid skeleton. The first terrestrial example of this class, spirojatamol (3), was recently discovered in root extracts of the roots of the Indian medicinal plant *Nardostachys jatamansi*.² Erythrodiene, the first marine hydrocarbon representative of this structure class, possesses the identical relative stereochemistry as that observed in spirojatamol.



Erythrodiene (1) was isolated by standard silica vacuum flash chromatography, and subsequent silica HPLC, from the condensed CH₂Cl₂ extracts of *E. caribaeorum* collected at Chub Cay, Bahamas, in June, 1989. Erythrodiene was analysed for C₁₅H₂₄ by combined HRMS and ¹³C NMR methods. The molecular formula and the observation

of four olefinic carbons in the ^{13}C NMR spectrum of erythrodiene, showed that the new sesquiterpene possessed two carbocyclic rings. The observed double bonds were assigned as terminal exocyclic olefins on the basis of their ^1H and ^{13}C NMR characteristics.³ In addition, ^1H NMR COSY experiments illustrated that two methyl doublets (δ 0.89 and 0.86), each showing $J = 7$ Hz couplings, were coupled to a one-proton multiplet at δ 1.45. On the basis of this observation, erythrodiene possessed an isopropyl group.

The spectral data, although clear, did not suggest that erythrodiene belonged to a common sesquiterpenoid group. In an attempt to prepare a crystalline derivative, erythrodiene was ozonized and the resulting crude ketone mix reduced with NaBH_4 in methanol to yield the crystalline diol 2. The diol showed typical spectral characteristics,⁴ but failed to yield significant structural data from its NMR spectra. However, since diol 2 was crystalline, the structure was solved by X-ray methods.



Diol 2 crystallized from CHCl_3 in space group $P2_12_12_1$ with $a = 6.385(1)$, $b = 10.908(2)$, $c = 17.987(3)$ and one molecule of $\text{C}_{13}\text{H}_{24}\text{O}_2$ per asymmetric unit. A total of 1014 independent reflections were measured using $\text{Cu K}\alpha$ and 2θ scans between $0^\circ \leq 2\theta \leq 116^\circ$, and 1008 (99%) were judged observed ($|F_o| \geq 3\sigma(|F_o|)$). The structure was successfully solved and refined with anisotropic heavy atoms and riding hydrogens to a final R-factor of 7.5%. Coordinates have been deposited with the Cambridge Crystallographic Data Centre.

Figure. Computer-generated drawing of diol 2.

Acknowledgements

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References

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2. Bagchi, A., Oshima, Y., Hikino, H., *Tetrahedron*, **46**, 1523, (1990).
3. For erythrodiene (1): an oil, $[\alpha]_D = -30.8^\circ$ (c 0.24, CHCl_3); LREIMS m/z 204 (M^+), 189 (23), 161 (66), 133 (56), 105 (57), 97 (75), 91 (85), 83 (76), 79 (50), 71 (81), 69 (89), 57 (100), 43 (65); HREIMS: $\text{M}^+ m/z = 204.1851$ calc. for $\text{C}_{15}\text{H}_{24}$ 204.1879. ^1H NMR (360 MHz, CDCl_3) δ 4.97 (s, 1H), 4.87 (s, 1H), 4.79 (s, 1H), 4.75 (s, 1H), 2.40 (m, 2-3H), 2.28 (m), 2.1 (m), 1.8 (m), 1.7 (m), 1.6 (m), 1.45 (m), 1.10 (m), 0.92 (m, 1H), 0.89 (d, $J = 7$ Hz, 3H), 0.85 (d, $J = 7$ Hz, 3H); ^{13}C NMR (50 MHz; CDCl_3): δ 158.0 C, 152.8 C, 106.7 CH₂, 109.9 CH₂, 51.3 C, 41.1 C, 39.7 CH, 39.6 CH₂, 33.8 CH₂, 33.0 CH₂, 32.5 CH, 31.1 CH₂, 20.8 CH₃, 20.1 CH₃, 19.7 CH₃.
4. For synthetic diol 2: IR (film) 3400, 2959, 2980, 1750, 1450, 1370 cm^{-1} ; LREIMS m/z 194, $\text{M}^+ - \text{H}_2\text{O}$, 176, $\text{M}^+ - 2\text{H}_2\text{O}$, 133, 107, 91, 67 amu; ^1H NMR (360 MHz, CDCl_3): δ 4.30 (m, 1H), 3.49 (dd, 1H), 1.4 - 2.1 (m, 14 H), 0.92 (d, $J = 7$ Hz, 3H), 0.88 (d, $J = 7$ Hz).

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